

Organocatalytic Enantioselective Michael–Michael–Michael–Aldol Condensation Reactions: Control of Five Stereocenters in a Quadruple-Cascade Asymmetric Synthesis of Highly Functionalized Hexahydrophenanthrenes

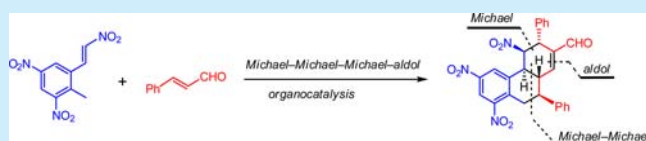
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S Supporting Information

ABSTRACT: A cascade organocatalysis has been developed for the enantioselective synthesis of a highly functionalized hexahydrophenanthrene-2-carbaldehyde containing five contiguous stereogenic centers with high diastereoselectivity and high enantioselectivity (>99% *ee*). The one-pot method comprises a cascade of organocatalytic Michael–Michael–Michael–aldol reactions of 2-methyl-1,5-dinitro-3-((*E*)-2-nitrovinyl)benzene and α,β -unsaturated aldehydes (e.g., cinnamaldehyde). The structure and absolute configuration of a product were confirmed by X-ray analysis of an appropriate derivative.



Cascade/domino organocatalysis reactions have attracted much attention during the past decade in the contexts of green chemistry and chemical efficiency;¹ consequently, this method has emerged as a pivotal strategy in contemporary organic synthesis.² Particularly, the ability to construct complex molecular architectures can be easily accessed via transformations beyond double cascade reactions. The pioneering *tour de force* involving triple cascade organocatalysis was introduced by Enders et al.³ However, fewer examples have been reported that involve quadruple cascade reactions that construct four new C–C bonds and that generate multiple stereocenters.^{4,5} Among those examples, the reactions were frequently triggered by a hetero-Michael or enol alkylation reaction. In addition, many double or triple cascade reactions have been initiated by oxa-Michael,⁶ sulfa-Michael,⁷ and aza-Michael⁸ reactions.⁹ The method of cascade organocatalysis initiated by a carbo-Michael reaction, especially with a benzylic nucleophile, toward an enal is rare and challenging, as compared to their heterocycle counterparts.¹⁰ Therefore, the development of exquisite higher-order cascade reactions in the construction of the aforementioned polycarbocycles is highly attractive and in great demand by synthetic chemists. Notably, hexahydrophenanthrene constitutes the basic skeleton of many naturally occurring and synthetic compounds possessing a number of interesting biological properties, and the hexahydrophenanthrene derivatives have received much attention in synthetic endeavors (Figure 1).

Prompted by the aforementioned background and in an effort to expand our exploration on organocatalyzed annulations,^{11,12} we envisioned that a quadruple cascade organocatalysis¹³ could afford a highly functionalized polycarbocyclic system with five contiguous stereogenic centers (Scheme 1). For example, retrosynthetic disconnection of hexahydrophenanthrene-2-

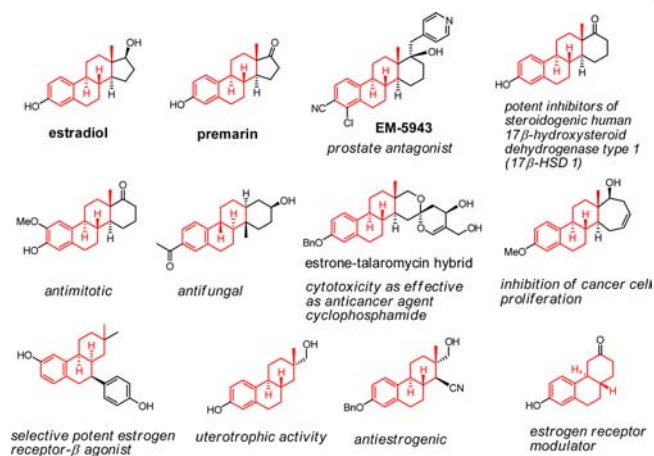


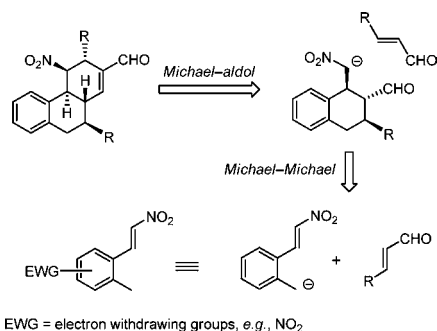
Figure 1. Selected examples of biologically active natural or synthetic compounds with the hexahydrophenanthrene core.

carbaldehyde via aldol and Michael transforms led to the 1-(nitromethyl)naphthalene-2-carbaldehyde and enal (Scheme 1). A subsequent double Michael transform of the adduct would yield the 1-methyl-2-((*E*)-2-nitrovinyl)benzene and enal. The methyl group on the toluenyl moiety can be activated by introducing electron-withdrawing groups, e.g., NO₂,¹⁴ on the aromatic ring. Therefore, the toluene derivative would be able to be deprotonated by a weak base and serve as an effective phenylogous nucleophile for the conjugate addition. Herein, we report the details of such an approach and the method that permitted enantioselective synthesis of a hexahydrophenanth-

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Scheme 1. Retrosynthetic Analysis



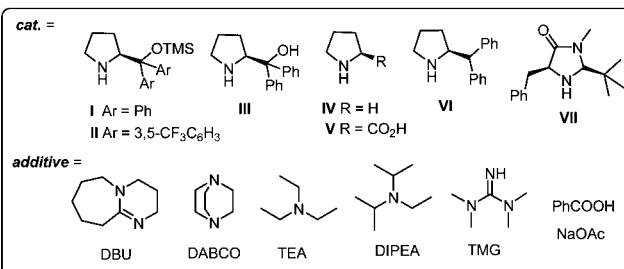
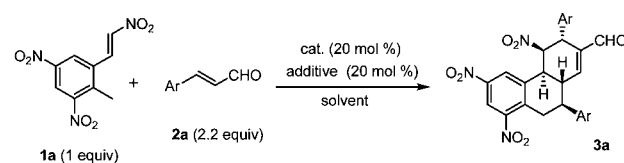
threne-2-carbaldehyde with five contiguous centers by domino carbo-Michael–Michael–Michael–aldol condensation.

At the outset of the study, the cascade reaction of (*E*)-2-methyl-1,5-dinitro-3-(2-nitrovinyl)benzene (**1a**)¹⁵ and cinnamaldehyde (**2a**) was performed with 20 mol % of Jørgensen–Hayashi catalyst (**I**)¹⁶ in ClCH₂CH₂Cl at ambient temperature for 5 days; this process only yielded a trace amount of the product (Table 1, entry 1). The same reaction conditions with the addition of benzoic acid and catalyst **I** also gave a similar result (Table 1, entry 2). In order to trigger the reaction of **1a**, the additive was replaced by a base, e.g., DBU, and to our delight, the reaction proceeded and gave 12% yield of hexahydrophenanthrene-2-carbaldehyde **3a** as the only observable diastereomer (Table 1, entry 3). Accordingly, the reaction was screened with a variety of base additives and catalyst **I** to obtain the optimum yield of adduct **3a** (Table 1, entries 4–7). With strong bases employed, e.g., DBU, DABCO, the reactions are readily reversible with quantitative recovery of the starting materials and polar residues. The best result was obtained in the reaction with the combination of diisopropylethyl amine (DIPEA, Hünig base) and catalyst **I** in ClCH₂CH₂Cl at ambient temperature for 60 h, affording a 25% yield of the adduct **3a** as the only observable diastereomer (Table 1, entry 7).

The same reaction in various solvents was screened for optimization with *cat.* **I**–DIPEA (20 mol %), e.g., toluene, CH₂Cl₂, CHCl₃, CH₃CN, EtOH, and THF (Table 1, entries 8–13). Apparently, the reaction in toluene provided the highest yield of product **3a**, 54%, along with the enantioselectivity >99% *ee* (Table 1, entry 8). On the other hand, the reaction with a strong base, 1,1,3,3-tetramethylguanidine (TMG), in THF only yielded a trace amount of the product **3a** (Table 1, entry 14). The reaction with an excess amount of additive DIPEA (100 mol %) and **I** (20 mol %) in toluene did not improve the yield but afforded less yield, 38% (Table 1, entry 15). The reaction was further scrutinized with a variety of organocatalysts (**II**–**VII**) along with DIPEA; however, none of those attempts gave promising results (Table 1, entries 16–21).

After securing the optimal reaction conditions (Table 1, entry 8), we next scrutinized the scope and limitations of the quadruple cascade organocatalysis reaction with variants of nitroalkene **1** and enal **2**. The reaction appeared quite general with respect to the substrates tested, providing the corresponding adducts with excellent stereoselectivities and in good yields (Table 2, entries 1–10). Usually, the reactions were completed within 3 days, but a longer reaction time was required in the case with chlorosubstituted nitroalkene **1** (Table 2, entries 9 and 10). Moreover, an additional amount of DIPEA (50 mol %) was needed to complete the reaction with **1i** and **1j**. It is noteworthy that although the cascade reaction providing product **3a** would

Table 1. Screening of the Catalysts, Additives, Solvents, and Reaction Conditions for the Cascade Reactions^a



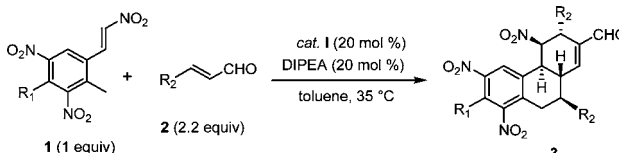
| entry | cat.–additive | solvent | time (h) | yield ^b (%) | ee ^d (%) |
|-------|----------------------|--------------------------------------|----------|------------------------|---------------------|
| 1 | I– | ClCH ₂ CH ₂ Cl | 120 | <5 ^c | nd |
| 2 | I–PhCOOH | ClCH ₂ CH ₂ Cl | 120 | <5 ^c | nd |
| 3 | I–DBU | ClCH ₂ CH ₂ Cl | 120 | 12 | nd |
| 4 | I–DABCO | ClCH ₂ CH ₂ Cl | 168 | 10 | nd |
| 5 | I–TEA | ClCH ₂ CH ₂ Cl | 48 | 23 | >99 |
| 6 | I–NaOAc | ClCH ₂ CH ₂ Cl | 120 | <5 ^c | nd |
| 7 | I–DIPEA | ClCH ₂ CH ₂ Cl | 60 | 25 | nd |
| 8 | I–DIPEA | toluene | 65 | 54 | >99 |
| 9 | I–DIPEA | CH ₂ Cl ₂ | 72 | 24 | >99 |
| 10 | I–DIPEA | CHCl ₃ | 96 | 21 | >99 |
| 11 | I–DIPEA | CH ₃ CN | 72 | nr ^e | nd |
| 12 | I–DIPEA | EtOH | 72 | nr ^e | nd |
| 13 | I–DIPEA | THF | 96 | 34 | 99 |
| 14 | I–TMG | THF | 96 | <5 ^c | nd |
| 15 | I–DIPEA ^c | toluene | 120 | 38 | nd |
| 16 | II–DIPEA | toluene | 168 | <5 ^c | nd |
| 17 | III–DIPEA | toluene | 168 | nr ^e | nd |
| 18 | IV–DIPEA | toluene | 96 | <5 ^c | 0 |
| 19 | V–DIPEA | toluene | 168 | nr ^e | na |
| 20 | VI–DIPEA | toluene | 96 | 22 | >99 |
| 21 | VII–DIPEA | toluene | 120 | nr ^e | na |

^aUnless otherwise noted, the reactions were performed with catalyst–additive (20 mol %) in 0.2 M of **1** with a ratio of 1/2.2 of **1a** and **2a** at ambient temperature. ^bIsolated yields of **3a**; dr >20:1. ^cRecovered most starting materials (**1a** and **2a**). ^dDetermined by HPLC with a chiral column (Chiralpak IC). ^e1.0 equiv of DIPEA (100 mol %) was added. nr = no reaction. nd = not determined. na = not available.

generate five new chiral centers, leading to a maximum of 32 possible stereoisomers, only one enantiomer was observed in this reaction. In addition, the overall yields of these quadruple cascade organocatalyses were ca. 50%, which is an average 85% yield for each C–C bond formation reaction. Moreover, except for the polar residues, the reactions are rather clean according to TLC and the crude ¹H NMR spectrum. Adduct **3** is the only isolable product in these reactions. The other side product (isomer or intermediate), if any, occurred in <5% yield.

The structure and the absolute configuration of the products were assigned based on X-ray analysis of (+)-**4i** (Figure 2), prepared from the reduction of **3i** (DIBAL-H, CH₂Cl₂, 0 °C, 1 h) and the subsequent esterification (4-Cl-C₆H₄COCl, Et₃N, *cat.* DMAP, CH₂Cl₂, 0 °C to rt, 2 h; 55% overall yield for two steps).

Table 2. Scope of the Asymmetric Quadruple Cascade Reaction of 1 and 2^a



| entry | R ₁ , R ₂ | time (h) | yield ^b (%) | ee ^c (%) |
|-----------------|--|----------|------------------------|---------------------|
| 1 | 3a R ₁ = H; R ₂ = C ₆ H ₅ | 65 | 54 | >99 |
| 2 | 3b R ₁ = H; R ₂ = 4-BrC ₆ H ₄ | 80 | 50 | >99 |
| 3 | 3c R ₁ = H; R ₂ = 4-MeC ₆ H ₄ | 60 | 61 | >99 |
| 4 | 3d R ₁ = H; R ₂ = 4-OMeC ₆ H ₄ | 84 | 48 | >99 |
| 5 | 3e R ₁ = H; R ₂ = 4-ClC ₆ H ₄ | 65 | 53 | >99 |
| 6 | 3f R ₁ = H; R ₂ = 4-FC ₆ H ₄ | 72 | 52 | >99 |
| 7 | 3g R ₁ = H; R ₂ = 3-MeC ₆ H ₄ | 72 | 42 | >99 |
| 8 | 3h R ₁ = H; R ₂ = 4-EtC ₆ H ₄ | 60 | 56 | >99 |
| 9 ^d | 3i R ₁ = 4-Cl; R ₂ = C ₆ H ₅ | 120 | 40 | >99 |
| 10 ^d | 3j R ₁ = 4-Cl; R ₂ = 4-MeC ₆ H ₄ | 144 | 42 | >99 |

^aUnless otherwise noted, the reactions were performed with catalyst I-additive (20 mol %) in 0.2 M of 1 with a ratio of 1/2.2 of 1a and 2a at 35 °C. ^bIsolated yields of 3a. ^cDetermined by HPLC with a chiral column (Chiralpak IC). ^d0.5 equiv of DIPEA was added.

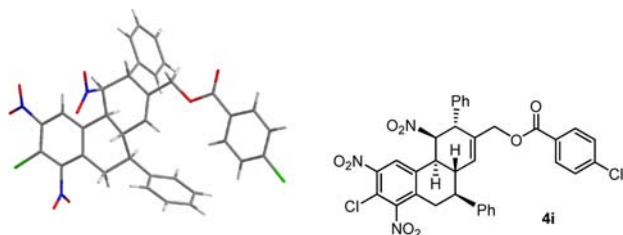
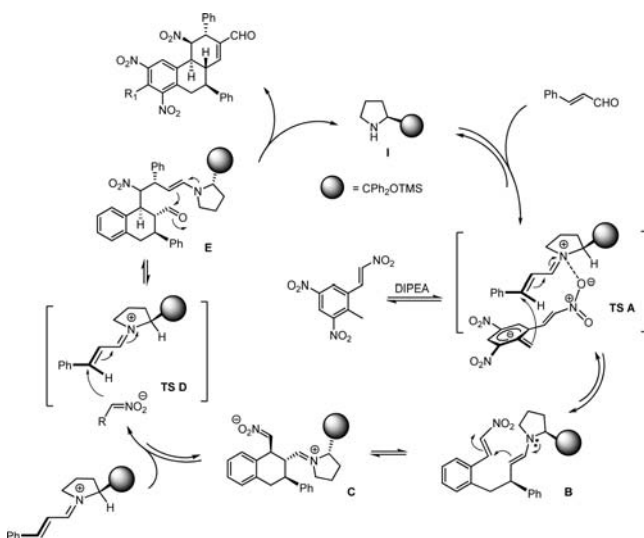


Figure 2. Stereoplots of the X-ray crystal structures of (+)-4i: C, gray; O, red; N, blue; Cl, green.

To account for the stereoselectivity of this transformation, we propose a plausible mechanism, as depicted in Scheme 2. Initial nucleophilic attack of a benzylic anion, generated *in situ* from 1a with DIPEA,¹⁷ on the iminium-activated cinnamaldehyde 2a from the *Re* face under the control of the catalyst (TS A) gives

Scheme 2. Plausible Reaction Mechanism



intermediate enamine B, which cyclizes to iminium C and subsequently reacts with another iminium-activated cinnamaldehyde 2a from the *Re* face under the control of the catalyst (TS D) to afford intermediate E. Consequently, the intramolecular aldol condensation of the intermediate provides 3a and regenerates the catalyst. The intriguing and excellent stereoselectivity of the cascade reaction may arise from synergistic effects: (1) the severe steric hindrance conferred by the bulky substituent at the catalyst, which hampers the reaction at the *Si* face (TS A); (2) the positive charge of the iminium ion on the pyrrolidinium to interact with the negative charge of the nitro substituent in 1a, acting as an “electron sink”.¹⁸ This dual effect directs the first-step Michael addition of the phenylogous dinitro toluene toward the iminium-activated 2a as depicted in Scheme 2 to afford excellent stereoselectivity.

In summary, we have realized a concise synthesis of optically enriched and highly functionalized hexahydrophenanthrenes, containing five contiguous stereogenic centers with excellent diastereo- (>20:1) and enantioselectivities (up to >99% ee) by quadruple cascade organocatalysis: triple-Michael/aldol reactions. Particularly noteworthy is the one-pot operation of the quadruple cascade reaction. The method not only adds to the limited repertoire of quadruple cascade organocatalytic asymmetric reactions but also demonstrates a proof of principle of the synergistic action of a steric shielding and an electron-sink mode of catalysis and nucleophilicity. The method has achieved cascade reactions that are hardly otherwise catalyzed by enamine catalysts alone, but require the addition of DIPEA. The success in triggering the first step cascade reactions by a carbo-Michael reaction of a benzylic nucleophile toward the enal further demonstrates the merit of this model and expands the realm of their synthetic applications. The structures and the absolute configuration of the products were unambiguously confirmed by single crystal X-ray crystallographic analyses of an appropriate adduct. Further applications of this protocol in the synthesis of elaborated derivatives, e.g., steroids, are currently underway.

■ ASSOCIATED CONTENT

Supporting Information

Experimental procedures and characterization data for the new compounds and X-ray crystallographic data for compound (+)-4i. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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(18) For a similar example of synergistic effects, see ref 10c.